

INDIAN RUBBER INSTITUTE
PGD-IRI EXAMINATION – 2017

Paper - I

Date : 19.08.2017

Time : 10.00-13.00 hrs.

Duration : 3 Hours

Full Marks : 100

POLYMER SCIENCE

Answers should be illustrated with sketches wherever helpful

Question number 1 is compulsory. Answer **FOUR** from the remaining questions taking **TWO** from each group

GROUP – A

1. Multiple choice questions. Select the correct answer from the given alternatives :
- (i) Shellac is a
a) Natural resin b) Synthetic resin c) Regenerated resin d) Rubber
- (ii) When the T_g of the polymer is well below room temperature, it is a
a) Plastic b) Rubber c) Fibre d) Dendrimer
- (iii) Polyethylene terephthalate is a
a) Polyamide b) Polyimide c) Polyether d) Polyester
- (iv) EVA is a
a) Homopolymer b) Copolymer c) Terpolymer d) non polymeric material
- (v) The weight average molecular weight is determined by
a) Spectroscopy b) Osmometry c) Viscometry d) Light scattering
- (vi) For a Hookean region stress is directly proportional to
a) Viscosity b) Strain rate c) Strain d) Modulus
- (vii) Aramids are high performance
a) Polyesters b) Polyethers c) Polyimides d) Polyamides
- (viii) NBR is a
a) Random copolymer b) Alternate copolymer
c) Block copolymer d) none of the above
- (ix) Polypropylene is known as
a) Glassy polymer b) Amorphous polymer c) Crystalline polymer d) Ionomer
- (x) Glass transition temperature of polystyrene is
a) -100°C b) -70°C c) $+100^\circ\text{C}$ d) $+150^\circ\text{C}$ [TURN OVER]

- (xi) Glass transition temperature of a polymer is determined by
a) Infrared spectrophotometer b) Differential scanning calorimeter
c) Mass spectrometer d) Scanning electron microscopy
- (xii) Polymer produced from anionic polymerization is called
a) Dead polymer b) Living polymer c) Ladder polymer d) none of these
- (xiii) Atactic polymers are
a) Crystalline b) Semi-crystalline c) Amorphous d) None of these
- (xiv) Copolymer of styrene and butadiene, which is used in tire is
a) SBS b) SBR c) SEBS d) XSBR
- (xv) IIR is commercially prepared by
a) Radical polymerization b) Ziegler – Natta polymerization
c) Cationic polymerization d) Anionic polymerization
- (xvi) Crystallinity of a polymer is quantitatively determined by
a) XRD b) TGA c) DETA d) DMA
- (xvii) If PE has a molecular weight (M_n) of 2,80,000 what is its degree of polymerization?
a) 100 b) 1000 c) 10000 d) None of the these
- (xviii) Name of a plasticizer for PVC is
a) DRC b) DCP c) DPG d) DBP
- (xix) Gutta Percha is
a) Cis 1,4 polyisoprene b) Trans 1,4 polyisoprene
c) Vulcanised polyisoprene d) Ebonite
- (xx) A rubber having good biocompatibility is
a) NR b) SBR c) NBR d) PDMS

(1 x 20) = 20

- 2.(a) Write down the different steps of free radical polymerization by taking styrene as **monomer** and by using AIBN as initiator.
- (b) How can you prepare Nylon 6 and Nylon 66?
- (c) Explain the term 'back biting mechanism'.
- (d) What is Ziegler – Natta Catalyst?

(8+6+4+2) = 20

- 3.(a) Write down the catalyst or initiator generally used to polymerize the following **monomers** and write down the structures of the resultant polymer and copolymer.
- i) Acrylonitrile and Butadiene
- ii) Isobutylene and Is

- iii) Phenol and Formaldehyde
- iv) Ethylene, propylene and 1,4 hexadiene.
- v) Terephthalic acid and Ethylene glycol.

5 x 2 = 10

- (b) Give the name of the method of preparation of the following polymers mentioning the corresponding monomers used.

- i) SBR ii) PU iii) S-B-S iv) BR v) FKM

5 x 2 = 10

- 4.(a) Explain the difference between suspension and emulsion polymerization.
(b) Explain the difference between cationic and anionic polymerization.
(c) Explain the difference between an initiator and an inhibitor.

(10+6+4) = 20

GROUP – B

5. Discuss the differences between
- (a) Newtonian fluid and Non-Newtonian fluid.
 - (b) Maxwell model and Voigt model.
 - (c) Stress relaxation and Creep.
 - (d) Pseudoplastic and dilatant material.
 - (e) Elasticity and viscosity.

(5 x 4) = 20

- 6.(a) Give a comparative account of the stress – strain plots for:
- (i) A ductile plastic.
 - (ii) A typical rigid / brittle plastic.
 - (iii) A typical fibre.
 - (iv) A rubber having strain induced crystallization.

- (b) What is 'hysteresis'? Explain its importance.
- (c) Explain the term 'shape factor'. What is its importance?
- (d) Define the term 'transmissibility'.

(8+5+5+2) = 20

7. Name the synthetic rubbers used in each case with the structure of the corresponding monomer and polymer in each case.

- (a) A rubber which exhibits very good low temperature flexibility.
- (b) A rubber which exhibits excellent weather resistant property.
- (c) A heat and oil resistant rubber.
- (d) A rubber used in Tyre tread compound.
- (e) A rubber used for making adhesive.

(1+3) x 5 = 20

[TURN OVER]

8. Write short notes on Any Four of the following

- (a) Z-average molecular weight
- (b) Polydispersity index
- (c) Dilatometer
- (d) Strain induced crystallization
- (e) Solubility parameter
- (f) Spherulite

(4 x 5) = 20